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Intellectual Property Status: Patent Pending

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Bicarbonate and Carbonate as Hydroxide Carriers in a Biological Fuel Cell

AzTE Case # M08-068

Invention Description

Biological fuel cells (BFCs) are being widely researched these days as a means to produce combustion-less electrical energy from organic compounds present in waste water. BFCs are capable of oxidizing many complex and simple organic compounds.

BFCs open up the possibility of producing electrical energy directly from biomass feed stocks that are renewable and carbon-neutral fuels. Currently, however, this technology is limited by the development of a pH gradient across the anode/cathode compartments of the cell that results in approximately a 20% loss in available energy output.

Researchers at the Biodesign Institute of Arizona State University have discovered new methods and devices for reducing the pH gradient between the cathode and anode compartments in fuel cells by supplying carbon dioxide to the cathode compartment. The gas flow to the cathode is controlled and maintained at a sufficient concentration to reduce the pH gradient and raise the energy output efficiency.

The technology can improve the performance of BFCS and make them a more viable source of renewable energy.

Potential Applications

- Alternative energy source
- Renewable energy source

Benefits and Advantages

- Improves biological fuel cell performance by increasing power output
- Carbon dioxide produced in the anode chamber can be recycled to the cathode chamber